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greater, since taking the calorimeter's heat capacity into account will compensate for the thermal energy transferred to the solution from the calorimeter; this approach includes the calorimeter itself, along with the solution, as "surroundings":  $q_{rxn} = -(q_{solution} + q_{calorimeter})$ ; since both  $q_{solution}$  and  $q_{calorimeter}$  are negative, including the latter term ( $q_{rxn}$ ) will yield a ...

### Answer Key Chapter 5 - Chemistry | OpenStax

5.E: Homework Chapter 5 Answers. 3. H + for hydrogen ion; H - for hydride ion. 5. a. Nitrogen,1; Hydrogen,4; Chlorine,1. b. Sodium,1; Carbon,1; Nitrogen,1. c. Barium,1; Hydrogen,2; Carbon,2; Oxygen,6.

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CHAPTER 5 REVIEW The Periodic Law SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. c When an electron is added to a neutral atom, energy is (a) always absorbed. (c) either absorbed or released. (b) always released. (d) neither absorbed nor released. 2. d The energy required to remove an electron from a neutral atom is the atom's

### 5 The Periodic Law

Identify the letter of the choice that best completes the statement or answers the question. \_\_\_\_ 1. The idea of arranging the elements in the periodic table according to their chemical and physical properties is attributed to. a. Mendeleev. c. Bohr. .... Chemistry Chapter 5 Exam ...

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Chemistry (4th Edition) Burdge, Julia Publisher McGraw-Hill Publishing Company ISBN 978-0-07802-152-7

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